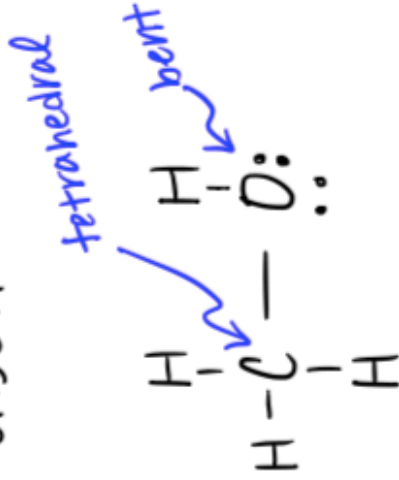


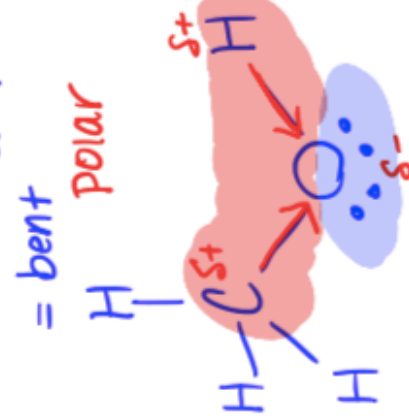
CH₃OH



Around C: 4 bonds
NO LP

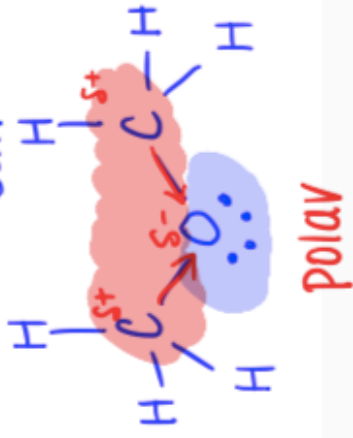
= tetrahedral

Around O: 2 bonds
2 LP

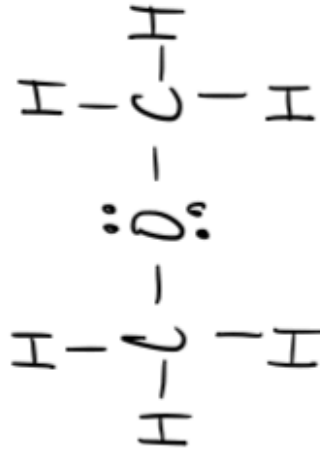


Around C: 4 bonds
= tetrahedral

Around O: 2 bonds
2 LP
= bent

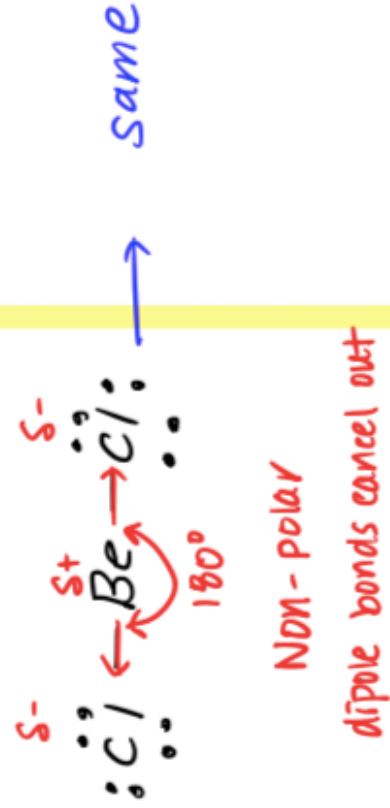


CH₃OCH₃



BeCl₂

Be: 2 bonds
= linear



Around C: 3 bonds
NO LP
= trigonal planar

